

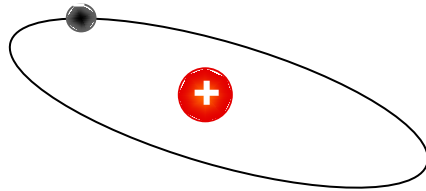
# Les particules élémentaires – Comment elles constituent l'atome

 **Proton**  
(charge électrique positive)

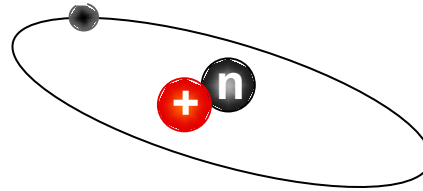
 **Neutron**  
(électriquement neutre)

 **Electron**  
(charge électrique négative)

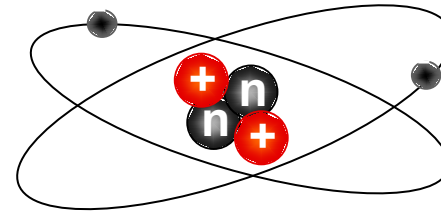
## Divers atomes



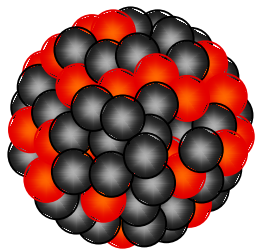
Hydrogène



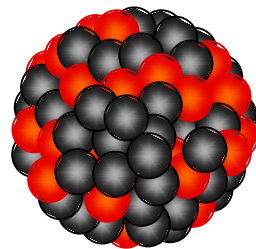
Hydrogène lourd



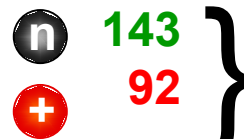
Hélium



<sup>238</sup><sub>92</sub>U



<sup>235</sup><sub>92</sub>U



} 235 **Nombre de masse:**  
nombre total de protons  
et de neutrons

Numéro atomique = nombre de protons